

Today's Objective:

- Identify the type of macromolecule(s) that make up enzymes
- Describe how enzymes function & the importance of enzymes in living organisms
- Given a graph, identify the optimal conditions for a particular enzyme's activity



<u>Chemical Reactions</u>

-<u>REACTANTS</u> (start with) \rightarrow <u>PRODUCTS</u> (end with) -Example: CH₄ + O₂ \rightarrow CO_{2(gas)} + H₂O





Activation Energy:

- All reactions need ENERGY to "GET STARTED"
- Definition: Activation energy barrier Energy required Activation energy to start a reaction Reactants Energy Energy Energy of reactants released Products Energy of products

Direction of reaction



Different activation energy's needed for different REACTIONS!





Factors That Speed Up Chemical Reactions

- I. High amount of reactants (lots to start with)
- 2. Higher temperatures (but not too hot!)
- 3. Neutral pH (not too acidic, not too basic!)



Enzymes in Biological Chemical Reactions

- -Most biochemical rxns <u>REQUIRE</u> enzymes
- -Why you ask?
 - I. Temp too low in cells/body
 - 2. Low amount of reactants in cells/body





- Enzymes:
 - All enzymes are large proteinsEnzymes act as CATALYSTS
- Catalysts:
 - –Speed up the rate chemical reactions \checkmark
 - -Decrease amount of activation

energy required

Enzymes lower the activation energy of a reaction!



Artist's rendition of the Magic Enzyme Fairy



Enzymes LOWER the ACTIVATION ENERGY of a reaction!







ENZYMES are Like Locks & Keys...Very Specific!

• Each chemical reaction has its own enzyme (each lock has its own key)





Affect of the Environment on Enzymes



Enzymes have optimal or "best" temp



Enzymes have optimal or "best" pH



(c) Reactant concentration

Enzyme activity increases with more reactant

Importance of Enzymes in Living Organisms

- -<u>4,000 biochemical reactions</u> catalyzed by enzymes!
- Regulate cell growth
- Allow movement
- Transport materials
- Make more DNA and then make proteins
- Digest macromolecules
- Build macromolecules
- And more!